

Academic Resource Sheet

CHEM 2400/2410

Class	General Formula	Example	Common Name (Systematic Name)	Common Suffix/Prefix (Systematic)
		Hydrocarbons		
Alkanes	RH	CH ₃ CH ₃	ethane	-ane
Alkenes	RR'C-CR"R"	H ₂ C=CH ₃	ethylene (ethene)	-ene
Alkynes	RC—CR'	нс-сн	acetylene (ethyne)	(-yne)
Arenes	ArH°	_ >	benzene	-ene
	Hai	logen-Containing Comp		
Alkyl halides	RX	CH ₃ CH ₂ CI	ethyl chloride (chloroethane)	halide (halo-)
Aryl halldes	ArX ^a		chlorobenzene	halo-
	Ox	ygen-Containing Comp	ounds	
Alcohols	ROH⁴	CH₃CH₂OH	ethyl alcohol (ethanol)	-ol
Phenols	ArOH ^b	ОН-ОН	phenol	-ol
Ethers	ROR'	H ₃ CH ₂ COCH ₂ CH ₃	diethyl ether	ether
Aldehydes	RCHO	CH³CH	acetaldehyde (ethanal)	-aldehyde (-al)
Setones	RR'C-O	СН₃ССН₃	acetone (2-propanone)	-one
Carboxylic acids	RCO ₂ H	O CH₃COH	acetic acid (ethanoic acid)	-ic acid (-oic acid)
	c	arboxylic Acid Derivath	ves	
sters	RCO₂R′	СН ₃ СОСН ₃	methyl acetate (methyl ethanoate)	-ate (-oate)
mides	RCONHR'	O ∥ CH₃CNHCH₃	N-methylacetamide	-amide
	Nitro	gen-Containing Compo	ounds	
mines	RNH ₂ , RNHR', RNR'R"	CH ₃ CH ₂ NH ₂	ethylamine	-amine
Itriles	RC≔N	H ₃ CC≡ N	acetonitrile	-nitrile
ltro compounds	ArNO ₂ °	NO ₂	nitrobenzene	nitro-

[&]quot;R indicates an alkyl group "Ar indicates an aryl group.

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Functional	_	pKa	Conjugate Base
Alkane	H ₃ C CH ₃	~50	H³C ∕_CH⁵⊝
Alkene	**	~43	9
Hydrogen	H-H	36	He
Amine	NH ₃	~35	NH ₂
Sulfoxide	O H₃C ^S CH₃	31	H₃C ^S ,CH₂
Alkyne	R——H	25	R——⊖
Ester	H ₃ CO CH ₃	25	H³CO CH⁵
Nitrile	H ₃ C _C N	25	H²C.C.N
Ketone/ aldehyde	H ₃ C CH ₃	20-24	H³C CH⁵
Alcohol	H₃C ^{OH}	17	H³C-O⊖
Water	но-н	16	но⊖
Malonates	H3CO H H OCH3	13 H	зсо ⊖ осн₃



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Thiols	CH₃S–H	13 CI	H₃S⊝
Protonated amines	NH₄ CI	9-11 _{NE}	13
Carboxylic acids	H³C OH	4 H₃C	الم⊝
Hydrofluoric acid	H-F	3.2 F	∋
Sulfonic acids Me'	(tosic acid)	-1 Ma)°\$.0⊖
Hydronium ion	H₃O ⊕	-1.7 H₂0)
Sulfuric acid	H ₂ SO ₄	3 HSC	, ,⊖
Hydrochloric acid	HCI	_6 CI [€]	
Hydrobromic acid	HBr	_9 Br €	
Hydrolodic acid	н	-10 I €)





